

of chlordiazepoxide ( $C_{16}H_{14}ClN_3O$ : 299.75).

**Method of preparation** Prepare as directed under Powder, with Chlordiazepoxide.

**Identification (1)** Weigh a portion of Chlordiazepoxide Powder, equivalent to 0.01 g of Chlordiazepoxide according to the labeled amount, add 100 mL of 0.1 mol/L hydrochloric acid TS, shake, and filter. To 5 mL of the filtrate add 0.1 mol/L hydrochloric acid TS to make 100 mL, and determine the absorption spectrum as directed under the Ultraviolet-visible Spectrophotometry: it exhibits maxima between 244 nm and 248 nm and between 306 nm and 310 nm, and a minimum between 288 nm and 292 nm.

(2) Weigh a portion of Chlordiazepoxide Powder, equivalent to 0.02 g of Chlordiazepoxide according to the labeled amount, add 10 mL of methanol, shake for 5 minutes, then filter by suction through a glass filter (G4), evaporate the filtrate with the aid of a current of air to dryness, and dry the residue in vacuum at 60°C for 1 hour. Determine the infrared absorption spectrum of the residue as directed in the potassium bromide disk method under the Infrared Spectrophotometry: it exhibits absorption at the wave numbers of about 1625  $cm^{-1}$ , 1465  $cm^{-1}$ , 1265  $cm^{-1}$ , 850  $cm^{-1}$  and 765  $cm^{-1}$ .

**Purity** Conduct this procedure without exposure to daylight, using light-resistant vessels. To a portion of Chlordiazepoxide Powder, equivalent to 0.050 g of Chlordiazepoxide according to the labeled amount, add exactly 5 mL of a mixture of methanol and ammonia TS (97:3), shake, centrifuge, and use the supernatant liquid as the sample solution. Separately, dissolve 0.050 g of Chlordiazepoxide Reference Standard in a mixture of methanol and ammonia TS (97:3) to make exactly 50 mL, and use this solution as the standard solution (1). Dissolve 5.0 mg of 2-amino-5-chlorobenzophenone for thin-layer chromatography in methanol to make exactly 200 mL, and use this solution as the standard solution (2). Perform the test with these solutions as directed under the Thin-layer Chromatography. Spot 25  $\mu L$  of the sample solution and 10  $\mu L$  each of the standard solutions (1) and (2) on a plate of silica gel with fluorescent indicator for thin-layer chromatography. Proceed as directed in the Purity (2) under Chlordiazepoxide.

**Assay** Conduct this procedure without exposure to daylight, using light-resistant vessels. Weigh accurately a quantity of Chlordiazepoxide Powder, equivalent to about 0.1 g of Chlordiazepoxide ( $C_{16}H_{14}ClN_3O$ ), transfer to a glass-stoppered flask, add exactly 100 mL of methanol, stopper, shake vigorously for 15 minutes, and centrifuge. Pipet 10 mL of the supernatant liquid, add exactly 5 mL of the internal standard solution, add methanol to make exactly 100 mL, and use this solution as the sample solution. Separately, weigh accurately about 0.01 g of Chlordiazepoxide Reference Standard, previously dried in a desiccator (in vacuum, phosphorus (V) oxide, 60°C) for 4 hours, dissolve in methanol, add exactly 5 mL of the internal standard solution, add methanol to make exactly 100 mL, and use this solution as the standard solution. Perform the test with 10  $\mu L$  each of the sample solution and the standard solution as directed under the Liquid Chromatography according to the following operating conditions, and calculate the ratios,  $Q_T$  and  $Q_S$ , of the peak area of chlordiazepoxide to that of the internal standard.

$$\begin{aligned} & \text{Amount (mg) of chlordiazepoxide (C}_{16}\text{H}_{14}\text{ClN}_3\text{O)} \\ &= \text{amount (mg) of Chlordiazepoxide} \\ & \text{Reference Standard} \\ & \times \frac{Q_T}{Q_S} \times 10 \end{aligned}$$

**Internal standard solution**—A solution of isobutyl salicylate in methanol (1 in 20).

**Operating conditions**—

**Detector:** An ultraviolet absorption photometer (wavelength: 254 nm).

**Column:** A stainless steel column about 4 mm in inside diameter and 25 to 30 cm in length, packed with octadecylsilanized silica gel for liquid chromatography (10  $\mu m$  in particle diameter).

**Column temperature:** A constant temperature of about 25°C.

**Mobile phase:** A mixture of methanol and 0.02 mol/L ammonium dihydrogenphosphate TS (7:3).

**Flow rate:** Adjust the flow rate so that the retention time of chlordiazepoxide is about 5 minutes.

**Selection of column:** Proceed with 10  $\mu L$  of the standard solution under the above operating conditions. Use a column giving elution of chlordiazepoxide and the internal standard in this order with the resolution between these peaks being not less than 9.

**Containers and storage** Containers—Tight containers.

## Chlordiazepoxide Tablets

クロルジアゼポキシド錠

Chlordiazepoxide Tablets contain not less than 93% and not more than 107% of the labeled amount of chlordiazepoxide ( $C_{16}H_{14}ClN_3O$ : 299.75).

**Method of preparation** Prepare as directed under Tablets, with Chlordiazepoxide.

**Identification (1)** Weigh a portion of powdered Chlordiazepoxide Tablets, equivalent to 0.01 g of Chlordiazepoxide according to the labeled amount, add 100 mL of 0.1 mol/L hydrochloric acid TS, shake, and filter. To 5 mL of the filtrate add 0.1 mol/L hydrochloric acid TS to make 100 mL, and determine the absorption spectrum as directed under the Ultraviolet-visible Spectrophotometry: it exhibits maxima between 244 nm and 248 nm and between 306 nm and 310 nm, and a minimum between 288 nm and 292 nm.

(2) Weigh a portion of powdered Chlordiazepoxide Tablets, equivalent to 0.01 g of Chlordiazepoxide according to the labeled amount, add 10 mL of diethyl ether, shake vigorously, and centrifuge. Evaporate 5 mL of the supernatant liquid by warming on a water bath to dryness. Determine the infrared absorption spectrum of the residue as directed in the potassium bromide disk method under the Infrared Spectrophotometry: it exhibits absorption at the wave numbers of about 1625  $cm^{-1}$ , 1465  $cm^{-1}$ , 1265  $cm^{-1}$ , 850  $cm^{-1}$  and 765  $cm^{-1}$ .

**Purity** Related substances—Conduct this procedure without exposure to daylight, using light-resistant vessels. To a portion of powdered Chlordiazepoxide Tablets, equivalent to 0.01 g of Chlordiazepoxide according to the labeled amount, add 10 mL of diethyl ether, shake vigorously, and centrifuge. Evaporate 5 mL of the supernatant liquid by warming on a water bath to dryness. Determine the infrared absorption spectrum of the residue as directed in the potassium bromide disk method under the Infrared Spectrophotometry: it exhibits absorption at the wave numbers of about 1625  $cm^{-1}$ , 1465  $cm^{-1}$ , 1265  $cm^{-1}$ , 850  $cm^{-1}$  and 765  $cm^{-1}$ .

lent to 0.050 g of Chlordiazepoxide according to the labeled amount, add exactly 5 mL of a mixture of methanol and ammonia TS (97:3), shake, centrifuge, and use the supernatant liquid as the sample solution. Separately, dissolve 0.050 g of Chlordiazepoxide Reference Standard in a mixture of methanol and ammonia TS (97:3) to make exactly 50 mL, and use this solution as the standard solution (1). Dissolve 5.0 mg of 2-amino-5-chlorobenzophenone for thin-layer chromatography in methanol to make exactly 200 mL, and use this solution as the standard solution (2). Perform the test with these solutions as directed under the Thin-layer Chromatography. Spot 25  $\mu$ L of the sample solution and 10  $\mu$ L each of the standard solutions (1) and (2) on a plate of silica gel with fluorescent indicator for thin-layer chromatography. Proceed as directed in the Purity (2) under Chlordiazepoxide.

**Dissolution test** Perform the test with 1 tablet of Chlordiazepoxide Tablets at 100 revolutions per minute according to Method 2 under the Dissolution Test, using 900 mL of diluted phosphate buffer solution, pH 6.8, (1 in 2) as the test solution. Take 30 mL or more of the dissolved solution 60 minutes after start of the test, and filter through a membrane filter with pore size of not more than 0.8  $\mu$ m. Discard the first 10 mL of the filtrate, pipet the subsequent  $V$  mL, add diluted phosphate buffer solution, pH 6.8, (1 in 2) to make exactly  $V'$  mL so that each mL contains about 3.7  $\mu$ g of chlordiazepoxide ( $C_{16}H_{14}ClN_3O$ ) according to the labeled amount, and use this solution as the sample solution. Separately, weigh accurately about 0.012 g of chlordiazepoxide for assay, previously dried in a desiccator for 4 hours (in vacuum, phosphorus (V) oxide, 60°C), and dissolve in diluted phosphate buffer solution, pH 6.8, (1 in 2) to make exactly 200 mL. Pipet 3 mL of this solution, add diluted phosphate buffer solution, pH 6.8, (1 in 2) to make exactly 50 mL, and use this solution as the standard solution. Determine the absorbances,  $A_T$  and  $A_S$ , of the sample solution and the standard solution at 260 nm as directed under the Ultraviolet-visible Spectrophotometry.

The dissolution rate of Chlordiazepoxide Tablets in 60 minutes is not less than 70%.

Dissolution rate (%) with respect to the labeled amount of chlordiazepoxide ( $C_{16}H_{14}ClN_3O$ )

$$= W_S \times \frac{A_T}{A_S} \times \frac{V'}{V} \times \frac{1}{C} \times 27$$

$W_S$ : Amount (mg) of chlordiazepoxide for assay.

$C$ : Labeled amount (mg) of chlordiazepoxide ( $C_{16}H_{14}ClN_3O$ ) in 1 tablet.

**Assay** Conduct this procedure without exposure to daylight, using light-resistant vessels. Weigh accurately a quantity of powdered Chlordiazepoxide Tablets, equivalent to about 0.1 g of Chlordiazepoxide ( $C_{16}H_{14}ClN_3O$ ), transfer to a glass-stoppered flask, add 10 mL of water, and shake well to disintegrate. Add 60 mL of methanol, shake well, add methanol to make exactly 100 mL, and centrifuge. Pipet 10 mL of the supernatant liquid, add exactly 5 mL of the internal standard solution, add methanol to make exactly 100 mL, and use this solution as the sample solution. Separately, weigh accurately about 0.01 g of Chlordiazepoxide Reference Standard, previously dried in a desiccator (in vacuum, phosphorus (V) oxide, 60°C) for 4 hours, dissolve in 1 mL of water and methanol, add exactly 5 mL of the internal

standard solution, add methanol to make exactly 100 mL, and use this solution as the standard solution. Perform the test with 10  $\mu$ L each of the sample solution and the standard solution as directed under the Liquid Chromatography according to the following operating conditions, and calculate the ratios,  $Q_T$  and  $Q_S$ , of the peak area of chlordiazepoxide to that of the internal standard.

$$\begin{aligned} &\text{Amount (mg) of chlordiazepoxide (C}_{16}\text{H}_{14}\text{ClN}_3\text{O)} \\ &= \text{amount (mg) of Chlordiazepoxide} \\ &\quad \text{Reference Standard} \\ &\quad \times \frac{Q_T}{Q_S} \times 10 \end{aligned}$$

**Internal standard solution**—A solution of isobutyl salicylate in methanol (1 in 20).

**Operating conditions**—

**Detector:** An ultraviolet absorption photometer (wavelength: 254 nm).

**Column:** A stainless steel column about 4 mm in inside diameter and 25 to 30 cm in length, packed with octadecylsilylated silica gel for liquid chromatography (10  $\mu$ m in particle diameter).

**Column temperature:** A constant temperature of about 25°C.

**Mobile phase:** A mixture of methanol and 0.02 mol/L ammonium dihydrogenphosphate TS (7:3).

**Flow rate:** Adjust the flow rate so that the retention time of chlordiazepoxide is about 5 minutes.

**Selection of column:** Proceed with 10  $\mu$ L of the standard solution under the above operating conditions. Use a column giving elution of chlordiazepoxide and the internal standard in this order with the resolution between these peaks being not less than 9.

**Containers and storage** Containers—Tight containers.

## Chlorhexidine Gluconate Solution

ゲルコン酸クロルヘキシジン液

Chlorhexidine Gluconate Solution is a solution of digluconate of chlorhexidine.

It contains not less than 19.0 w/v% and not more than 21.0 w/v% of chlorhexidine gluconate ( $C_{22}H_{30}Cl_2N_{10} \cdot 2C_6H_{12}O_7$ ; 897.76).

**Description** Chlorhexidine Gluconate Solution is a clear, colorless or pale yellow liquid. It is odorless, and has a bitter taste.

It is miscible with water and with acetic acid (100). 1 mL of Chlorhexidine Gluconate Solution is miscible with not more than 5 mL of ethanol (99.5) and with not more than 3 mL of acetone. By further addition of each of these solvents, a white turbidity is formed.

It is gradually colored by light.

Specific gravity  $d_{20}^{20}$ : 1.06 – 1.07

**Identification (1)** To 0.05 mL of Chlorhexidine Gluconate Solution add 5 mL of methanol, 1 mL of bromine TS and 1 mL of 8 mol/L sodium hydroxide TS: a deep red color is produced.

**(2)** To 0.5 mL of Chlorhexidine Gluconate Solution add