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Storage-Without fill up, light-resistant, remote from fire, and not exceeding 25°C.

Containers and storage Containers—Tight containers. Storage—Light-resistant.

## **Ethinylestradiol**

エチニルエストラジオール

C20H24O2: 296.40  $17\alpha$ -Ethynylestra-1,3,5(10)-triene-3,17 $\beta$ -diol [57-63-6]

Ethinylestradiol, when dried, contains not less than 98.0% of  $C_{20}H_{24}O_2$ .

Description Ethinylestradiol occurs as white to pale yellow crystals or crystalline powder. It is odorless.

It is freely soluble in pyridine and in tetrahydrofuran, soluble in ethanol (95) and in diethyl ether, and practically insoluble in water.

It dissolves in sodium hydroxide TS.

**Identification** (1) Dissolve 2 mg of Ethinylestradiol in 1 mL of a mixture of ethanol (95) and sulfuric acid (1:1): a purplish red color develops with a yellow-green fluorescence. Add carefully 2 mL of water to this solution: the color of the solution changes to red-purple.

(2) Transfer 0.02 g of Ethinylestradiol to a glass-stoppered test tube, dissolve in 10 mL of a solution of potassium hydroxide (1 in 20), add 0.1 g of benzoyl chloride, and shake. Collect the resulting precipitate, recrystallize from methanol, and dry in a desiccator (in vacuum, phosphorus (V) oxide): the precipitate melts between 200°C and 202°C.

**Optical rotation**  $[\alpha]_D^{20}$ :  $-26 - -31^{\circ}$  (after drying, 0.1 g, pyridine, 25 mL, 200 mm).

Melting point 180 – 186°C or 142 – 146°C

Purity Estrone—Dissolve 5 mg of Ethinylestradiol in 0.5 mL of ethanol (95), and add 0.05 g of 1,3-dinitrobenzene. Add 0.5 mL of freshly prepared dilute potassium hydroxide-ethanol TS, allow to stand in a dark place for 1 hour, and add 10 mL of ethanol (95): the solution has no more color than the following control solution.

Control solution: Proceed in the same manner as mentioned above, omitting Ethinylestradiol.

Loss on drying Not more than 0.5% (0.5 g, in vacuum, phosphorus (V) oxide, 4 hours).

**Residue on ignition** Not more than 0.1% (0.5 g).

Assay Weigh accurately about 0.2 g of Ethinylestradiol, previously dried, and dissolve in 40 mL of tetrahydrofuran. Add 10 mL of a solution of silver nitrate (1 in 20), and titrate with 0.1 mol/L sodium hydroxide VS (potentiometric titration).

> Each mL of 0.1 mol/L sodium hydroxide VS  $= 29.641 \text{ mg of } C_{20}H_{24}O_2$

## **Ethinylestradiol Tablets**

エチニルエストラジオール錠

Ethinylestradiol Tablets contain not less than 90% and not more than 110% of the labeled amount of ethinylestradiol ( $C_{20}H_{24}O_2$ : 296.40).

Method of preparation Prepare as directed under Tablets, with Ethinylestradiol.

Identification (1) Evaporate to dryness 5 mL of the sample solution obtained in Assay, and add 2 mL of a mixture of sulfuric acid and ethanol (95) (2:1) to the residue: a light red color with a yellow fluorescence develops. To the solution add carefully 4 mL of water: the color of the solution changes to red-purple.

(2) Evaporate to dryness 10 mL of the sample solution obtained in Assay, add 0.2 mL of acetic acid (31) and 2 mL of phosphoric acid to the residue, and heat on a water bath for 5 minutes: a red color with a yellow-green fluorescence develops.

Content uniformity Place 1 tablet of Ethinylestradiol Tablets in a separator, add 10 mL of the 2nd fluid of test fluids under the Disintegration Test, and shake until the tablet is disintegrated. Add 10 mL of dilute sulfuric acid and 20 mL of chloroform, shake vigorously for 5 minutes, and filter the chloroform layer into a conical flask through filter paper on which 5 g of anhydrous sodium sulfate is placed. Extract the aqueous layer with two 20-mL portions of chloroform, proceed with the extracts in the same manner as before, and combine the filtrates with the previous one. Evaporate gently the combined filtrate on a water bath with the aid of a current of nitrogen, dissolve the residue in exactly 100 mL of methanol, and centrifuge, if necessary. Pipet x mL of the supernatant liquid, add methanol to make exactly  $V \, \text{mL}$  of a solution containing about 0.04  $\mu g$  of ethinylestradiol (C<sub>20</sub>H<sub>24</sub>O<sub>2</sub>) per mL, and use this solution as the sample solution. Separately, weigh accurately about 0.01 g of Ethinylestradiol Reference Standard, previously dried in a desiccator (in vacuum, phosphorus (V) oxide) for 4 hours, dissolve in methanol, dilute to a volume containing about  $0.04 \,\mu g$  of ethinylestradiol ( $C_{20}H_{24}O_2$ ) per mL, and use this solution as the standard solution. Pipet 4 mL each of sulfuric acid-methanol TS into three glass-stoppered test tubes, T, S and B, cool in ice, to each tube add exactly 1 mL each of the sample solution, the standard solution and methanol, shake immediately, and allow to stand in a water bath at 30°C for 40 minutes, then allow to stand in a water bath at 20°C for 5 minutes. Perform the test with these solutions as directed under the Fluorometry. Determine the fluorescence intensities,  $F_T$ ,  $F_S$  and  $F_B$ , of these solutions using the fluorophotometer, at about 460 nm of the excitation and at about 493 nm of the fluorescence.

Amount (mg) of ethinylestradiol (C<sub>20</sub>H<sub>24</sub>O<sub>2</sub>)

= amount (mg) of Ethinylestradiol Reference Standard  $\times \frac{F_{\rm T} - F_{\rm B}}{F_{\rm S} - F_{\rm B}} \times \frac{V}{2500} \times \frac{1}{x}$ 

$$\times \frac{F_{\rm T} - F_{\rm B}}{F_{\rm S} - F_{\rm B}} \times \frac{V}{2500} \times \frac{1}{x}$$