Each mL of 0.1 mol/L perchloric acid VS  $= 28.380 \text{ mg of } C_{15}H_{21}NO_2.HCl$ 

Containers and storage Containers—Well-closed contain-

Storage—Light-resistant.

## **Indigocarmine**

インジゴカルミン

C<sub>16</sub>H<sub>8</sub>N<sub>2</sub>Na<sub>2</sub>O<sub>8</sub>S<sub>2</sub>: 466.35 Disodium 3,3'-dioxo-[△2.2'-biindoline]-5,5'-disulfonate [860-22-0]

Indigocarmine, when dried, contains not less than 95.0% of  $C_{16}H_8N_2Na_2O_8S_2$ .

Description Indigocarmine occurs as blue to dark blue powder or granules. It is odorless.

It is sparingly soluble in water, and practically insoluble in ethanol (95) and in diethyl ether.

It is hygroscopic.

When compressed, it has a coppery luster.

**Identification** (1) A solution of Indigocarmine (1 in 100) is dark blue in color. Perform the following tests with this solution as the sample solution: the dark blue color of each solution disappears.

- (i) Add 1 mL of nitric acid to 2 mL of the sample solution:
- Add 1 mL of bromine TS to 2 mL of the sample so-(ii) lution;
- (iii) Add 1 mL of chlorine TS to 2 mL of the sample solution;
- (iv) Add 2 mL of sodium hydroxide TS and 0.2 g of zinc powder to 2 mL of the sample solution, and warm.
- (2) Dissolve 0.1 g of Indigocarmine in 100 mL of a solution of ammonium acetate (1 in 650). To 1 mL of the solution add a solution of ammonium acetate (1 in 650) to make 100 mL. Determine the absorption spectrum of the solution as directed under the Ultraviolet-visible Spectrophotometry, and compare the spectrum with the Reference Spectrum: both spectra exhibit similar intensities of absorption at the same wavelengths.
- (3) Ignite 1 g of Indigocarmine to carbonize. After cooling, add 20 mL of water to the residue, shake, and filter the mixture: the filtrate responds to the Qualitative Tests for sodium salt and for sulfate.

pH Dissolve 0.10 g of Indigocarmine in 20 mL of water: the pH of the solution is between 5.0 and 6.0.

Purity (1) Water-insoluble substances—To 1.00 g of Indigocarmine add 200 mL of water, shake, and filter through a tared glass filter (G4). Wash the residue with water until the blue color of the filtrate becomes practically colorless, and dry the residue at 105°C for 4 hours: the mass of the residue does not exceed 5.0 mg.

(2) Arsenic—Place 0.8 g of Indigocarmine in a Kieldahl flask, add 5 mL of sulfuric acid and 5 mL of nitric acid, and ignite gently. Repeat the addition of 2 to 3 mL of nitric acid occasionally, and continue to heat until a colorless to light yellow solution is obtained. After cooling, add 15 mL of a saturated ammonium oxalate solution, heat the solution until dense white fumes are evolved, and concentrate to 2 to 3 mL. After cooling, dilute with water to 10 mL, and perform the test using Apparatus B with 5 mL of this solution as the test solution (not more than 5 ppm).

Loss on drying Not more than 10.0% (1 g, 105°C, 2 hours).

Residue on ignition Not less than 28.0% and not more than 38.0% (after drying, 1 g).

Assay Weigh accurately about 0.5 g of Indigocarmine, previously dried, add 15 g of sodium hydrogen tartrate monohydrate, and dissolve in 200 mL of water, boil with bubbling of a stream of carbon dioxide, and titrate, while being hot, with 0.1 mol/L titanium (III) chloride VS until the color of the solution changes from blue through yellow to orange.

Each mL of 0.1 mol/L titanium (III) chloride VS  $= 23.318 \text{ mg of } C_{16}H_8N_2Na_2O_8S_2$ 

Containers and storage Containers—Tight containers. Storage—Light-resistant.

## **Indigocarmine Injection**

インジゴカルミン注射液

Indigocarmine Injection is an aqueous solution for injection. It contains not less than 95% and not more than 105% of the labeled amount of indigocarmine  $(C_{16}H_8N_2Na_2O_8S_2: 466.35).$ 

Method of preparation Prepare as directed under Injection, with Indigocarmine.

**Description** Indigocarmine Injection is a dark blue liquid. pH: 3.0 - 5.0

Identification (1) To a volume of Indigocarmine Injection, equivalent to 0.02 g of Indigocarmine according to the labeled amount, add 1 mL of nitric acid: the dark blue color of the liquid disappears, and a yellow-brown color develops.

- (2) To a volume of Indigocarmine Injection, equivalent to 0.02 g of Indigocarmine according to the labeled amount, add 1 mL of bromine TS: the dark blue color disappears, and a yellow-brown color develops.
- (3) To a volume of Indigocarmine Injection, equivalent to 0.02 g of Indigocarmine according to the labeled amount, add 1 mL of chlorine TS: the dark blue color disappears, and a yellow-brown color develops.
- (4) To a volume of Indigocarmine Injection, equivalent to 0.01 g of Indigocarmine according to the labeled amount, add ammonium acetate solution (1 in 650) to make 1000 mL, and determine the absorbance of the solution as directed under the Ultraviolet-visible Spectrophotometry: it ex-